



Quantification of F₂-isoprostane isomers in cultured human lung epithelial cells after silica oxide and metal oxide nanoparticle treatment by liquid chromatography/tandem mass spectrometry

Xiaoqian Liu, Philip D. Whitefield, Yinfa Ma*

Department of Chemistry and Environmental Research Center, Missouri University of Science and Technology, Rolla, MO 65409, USA

ARTICLE INFO

Article history:

Received 18 December 2009
Received in revised form 4 March 2010
Accepted 8 March 2010
Available online 17 March 2010

Keywords:

F₂-isoprostane isomers
Oxidative stress
Nanoparticles
Solid phase extraction
Cell membrane
LC-MS/MS

ABSTRACT

F₂-isoprostanes are lipid peroxidation products of arachidonic acid in cell membrane and are reliable biomarkers for oxidative stress and cell membrane damage. Nanomaterials are widely used as raw materials in many industries and will have high potentials to be used in life science and medical fields. However, the human health impact of nanoparticles has caused people's great concern. Unfortunately, the mechanisms of cytotoxicity of many nanoparticles are not well defined. By measuring the levels of F₂-isoprostane isomers in cultured cells after nanoparticle exposure, the information can be used to explain whether the cytotoxicity of nanoparticles is caused by lipid peroxidation and to investigate the biological pathways. In this study, a liquid chromatography–tandem mass spectrometry (LC–MS/MS) method was developed to separate and quantify F₂-isoprostane isomers in nanoparticle-treated human lung cancer cells. Silica oxide (15 nm) and other four metal oxide nanoparticles including Fe₂O₃ (30 nm), Al₂O₃ (13 nm), TiO₂ (40 nm) and ZnO (70 nm) are chosen in this study. The isotope forms of F₂-isoprostane isomers, 8-iso-PGF_{2α}-d₄ and PGF_{2α}-d₄, were used as internal standard (IS). After human lung epithelial cells were exposed to different nanoparticles for 24 h, F₂-isoprostanes were extracted by a single step solid phase extraction with Oasis HLB cartridge. For the first time, six F₂-isoprostane isomers were tentatively identified and quantified in human lung epithelial cells. The levels of F₂-isoprostane isomers in the cells increased after the treatment with nanoparticles. For SiO₂, Fe₂O₃, and ZnO nanoparticles, F₂-isoprostane isomers increasing are consistent with nanoparticles' cytotoxicity data. For Al₂O₃ and TiO₂ nanoparticles, F₂-isoprostane isomers levels increased even before nanoparticles showed significant cytotoxicity at 100 μg/mL concentration in 24 h. Based on our best knowledge, this is the first study on the F₂-isoprostane isomers corresponding to nanoparticles' exposure *in vitro*. Our study demonstrates that SiO₂ (15 nm) nanoparticle showed the highest degree of lipid peroxidation and cell membrane damage among the studied nanoparticles.

© 2010 Elsevier B.V. All rights reserved.

1. Introduction

Nanotechnology has been developed for several decades depending on nanoparticles as its raw materials. The key features of nanoparticles are small particle size and large specific surface area [1]. Because of these special properties, nanoparticles have wide applications in catalysts, semiconductors, additives, and cosmetics [2–4]. Moreover, many researchers have used nanoparticles in biosensor, drug delivery and clinical treatment, especially for cancer treatment [5–9]. Recently, nanotoxicity, which means the

toxicity was caused by nanoparticles, has caused people's great concern [1,10,11]. Many studies, including our previous studies [12–17], have shown that nanoparticles are cytotoxic and can cause oxidative stress and DNA damage *in vitro* [18–20]. Our previous study reported that nanoparticles can cause lipid peroxidation and cell membrane damage by measuring total F₂-isoprostane levels as biomarkers [21].

F₂-isoprostanes are a series of compounds formed from free radical initiated lipid peroxidation of arachidonic acid in the cell membrane. They are reliable biomarkers for oxidative stress and lipid peroxidation. The proposed mechanism is that free radicals such as hydroxyl radicals can attack C₇, C₁₀ and C₁₃ three different sites of arachidonic acid carboxyl chain and abstract an allylic hydrogen. This process results in a delocalized pentadienyl carbon-centered radical. Subsequently, an oxygen molecule is inserted and a peroxy radical is formed. These peroxy radicals undergo further cyclization, followed by the addition of another oxygen molecule

* Corresponding author at: Department of Chemistry, Missouri University of Science and Technology, 400 West 11th Street, Rolla, MO 65409, USA.

Tel.: +1 573 341 6220; fax: +1 573 341 6033.

E-mail addresses: xlwv9@mst.edu (X. Liu), pwhite@mst.edu (P.D. Whitefield), yinfa@mst.edu (Y. Ma).

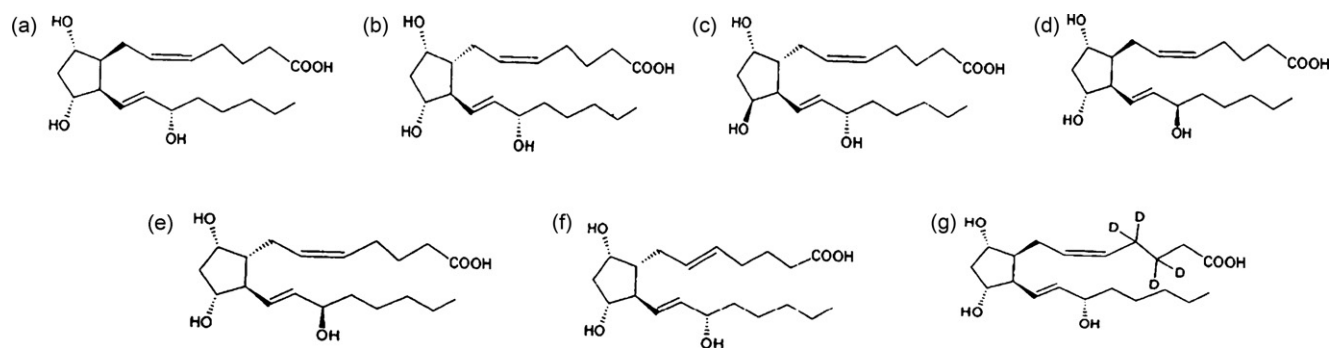


Fig. 1. Structures of F_2 -isoprostane regioisomers. (a) 8-Iso-PGF $_{2\alpha}$, (b) PGF $_{2\alpha}$, (c) 11 β -PGF $_{2\alpha}$, (d) 8-iso-15R-PGF $_{2\alpha}$, (e) 15R-PGF $_{2\alpha}$, (f) 5-trans-PGF $_{2\alpha}$, and (g) 8-iso-PGF $_{2\alpha}$ -d $_4$.

to yield bicyclic endoperoxide molecules. These intermediates are then reduced to F_2 -isoprostanes, named because they possess F-type prostane rings. In theory, four series (5-, 8-, 12-, 15-series) and 64 regioisomers of peroxidation products can be formed [22]. F_2 -isoprostanes are formed in situ as esterified form in phospholipids of cell membrane. The enzyme named phospholipase catalyzes the esterified F_2 -isoprostanes to free F_2 -isoprostanes and free F_2 -isoprostanes are released to biological fluids such as plasma and urine. In normal people, urinary F_2 -isoprostanes concentrations were 0.2–1.5 $\mu\text{g/g}$ creatinine [23]. It is important to study the isomers of F_2 -isoprostanes corresponding to oxidative injury because each isomer may have different biological functions. Moreover, it can help researchers to study the pathway of F_2 -isoprostanes formation in more detail. Among these isomers, 8-iso-PGF $_{2\alpha}$ has been widely studied and was found to possess some biological functions like nonspecific vasoconstriction, bronchoconstriction, and modulation of platelet function [24–26]. But other isomer's biological functions are not clear yet.

Gas chromatography/mass spectrometry (GC/MS) method has been used for measuring F_2 -isoprostanes. Even though the method is sensitive, the procedure is very tedious and time-consuming. It required multi-steps of solid phase extraction, thin layer chromatography, and derivatization reactions [27,28]. Enzyme immunoassay (ELISA) is commercial available. But it can cause cross-linking reaction, making the data not reliable. Moreover, each assay kit can only measure one isomer of F_2 -isoprostanes, making the assay not efficient as well. Liquid chromatography–tandem mass spectrometry (LC–MS/MS) method has been proven to be sensitive and selective for simultaneously measurement of different isomers of F_2 -isoprostanes without derivatizations in urine, plasma, and cell culture matrix [29–31]. In our previous study, total free F_2 -isoprostanes were determined by LC–MS method after the treatment of selected nanoparticles in human lung epithelial cells [21]. However, the changes of each F_2 -isoprostane isomer after nanoparticle treatments were not investigated. In this study, a one-cartridge solid phase extraction method using oasis HLB cartridge was further improved to efficiently extract F_2 -isoprostanes from complicated cell culture matrix, and a LC–MS/MS method was developed to separate and quantify six F_2 -isoprostane isomers in human lung epithelial cells after silica oxide and metal oxide nanoparticle exposure.

In this paper, five different nanoparticles were chosen for the study because they are more abundant in the environment and widely used in the industry. These nanoparticles include SiO $_2$ (15 nm), Fe $_2$ O $_3$ (30 nm), Al $_2$ O $_3$ (13 nm), TiO $_2$ (40 nm) and ZnO (70 nm). A549 human lung epithelial cell line (CCL-185, ATCC number) was used to study the exposure to nanoparticles because lung is the major organ affected when nanoparticles are inhaled. Six 15-series F_2 -isoprostane isomers were tentatively determined in the cell culture after cellular exposure to the nanoparticles. The molecular structures of these isomers are shown in Fig. 1. The levels of

F_2 -isoprostane isomers were quantified by LC–MS/MS method after exposure to each type of selected nanoparticles. The goal of this study was to explore and compare the response of different isomers and the extents of lipid peroxidation corresponding to nanoparticle treatment. This information can be very useful for revealing the mechanism of nanoparticle cytotoxicity and discover the biological functions of F_2 -isoprostane isomers.

2. Experimental

2.1. Chemicals

Authentic standards of 8-iso-PGF $_{2\alpha}$, 8-iso-15R-PGF $_{2\alpha}$, 5-trans-PGF $_{2\alpha}$, 11 β -PGF $_{2\alpha}$, 15R-PGF $_{2\alpha}$, PGF $_{2\alpha}$ and internal standards of 8-iso-PGF $_{2\alpha}$ -d $_4$, PGF $_{2\alpha}$ -d $_4$ were purchased from Cayman Chemical (Ann Arbor, MI, USA). HPLC/MS grade acetonitrile, methanol, HPLC grade hexane, ethyl acetate, and 2-propanol were purchased from Fisher Scientific (Pittsburgh, PA). Fetal bovine serum was purchased from American Type Culture Collection (ATCC) (Manassas, VA). Ham's F-12 medium with L-glutamine was purchased from Fisher Scientific (Pittsburgh, PA). Trypsin–EDTA (1 \times) was purchased from Invitrogen (Carlsbad, CA). HPLC/MS grade formic acid, penicillin–streptomycin, and citrate acid were obtained from Sigma–Aldrich (St. Louis, MO). Ultra pure deionized water was obtained from an Elix Milli-Q system (Millipore, Bedford, MA). All pH measurements were performed using an Accumet XL15 meter (Fisher Scientific).

2.2. Nanoparticles

The SiO $_2$ nanoparticle was purchased from Degussa Co. (Parsippany, NJ). The Fe $_2$ O $_3$ nanoparticle was purchased from Nanophase Technologies (Romeoville, IL). The ZnO nanoparticle was purchased from Sigma–Aldrich (St. Louis, MO). The TiO $_2$ nanoparticle was purchased from NanoScale Corporation (Manhattan, KS). The Al $_2$ O $_3$ nanoparticle was synthesized by the room temperature homogeneous nucleation method [32]. Transmission electron microscopy (TEM; Philips EM420) was used to measure particle size and distribution. The particle sizes and distributions of these nanoparticles were listed in Table 1. Characterization of nanoparticle has been studied previously from our group [18,19]. Nanoparticles tend to

Table 1
Nanoparticles size and distribution.

Nanoparticle	Size and distribution (mean \pm std) (nm)
SiO $_2$	15 \pm 5
Fe $_2$ O $_3$	30 \pm 5
Al $_2$ O $_3$	13 \pm 2
TiO $_2$	40 \pm 8
ZnO	70 \pm 13

aggregate in the solution over time. Therefore, nanoparticles were prepared freshly each time and dispersed by sonication for 15 min. Nanoparticle suspensions were applied to the cells immediately.

2.3. Assessment of nanoparticle cytotoxicity

The cells were treated with various dosages of nanoparticles. After exposure for 24–48 h, cell viability was measured by MTS assay (from Promega). The MTS assay was conducted as follows: under yellow light, 100 μ L of PMS (phenazine methosulfate) solution was added to 2.0 mL of MTS (3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium) solution immediately. 20 μ L of the combined MTS/PMS solution was added into each well of a 96-well plate. Each well contained 100 μ L of cells medium with 4000 cells. The plate was then incubated for 2 h at 37 °C in a humidified, 5% CO₂ incubator. Absorbance at 490 nm was recorded using a plate reader (FLUOstar OPTIMA).

2.4. Cell culture and treatment with nanoparticles

The human lung epithelial cell line (A549) was purchased from ATCC (Manassas, VA). This cell line was chosen because it is a typical cell line that has widely been used for *in vitro* cytotoxicity study [18,19]. The cells were grown in T-75 cell culture flasks filled with Ham's F-12 medium supplemented with 10% fetal bovine serum, 100 units/mL penicillin, and 100 μ g/mL streptomycin. Cells were grown at 37 °C in a 5% CO₂ humidified incubator, and were split every 2–3 days.

For the nanoparticle treatment experiment, cells were seeded in a 65 × 15 mm cell culture Petri dish at a density of 6 × 10⁵ cells/mL cell medium. Cells were allowed to attach to the Petri dish for 24 h. Nanoparticle suspensions in the cell medium were freshly prepared each time. Appropriate nanoparticles were weighed and added to the cell medium to make the stock solution. The suspensions were vortex and sonicated for 15 min. The stock solution was diluted for several steps to get the working solution; after each dilution the suspensions were vortex and sonicated for 15 min. Therefore the nanoparticles were evenly dispersed in the cell medium. The old cell medium was removed from the Petri dish, and the freshly prepared cell medium with nanoparticles was added to the cells. Cells filled with the culture medium without nanoparticles were served as the control. Cells were treated with nanoparticles for 24–48 h. After that, cell medium was collected. Antioxidant 1% 1N citrate acid was added to the cell medium. The samples were stored in –80 °C freezer immediately until analysis.

2.5. Solid phase extraction of F₂-isoprostanes

Free F₂-isoprostanes in the cell medium were extracted by Oasis HLB extraction cartridge (Waters, Milford, MA). 2 ng of internal standard 8-iso-PGF_{2 α} -d₄, PGF_{2 α} -d₄ was added to each sample. Then 3 mL of 2% formic acid was added to the samples to adjust the pH to 3. The cartridges were conditioned and equilibrated with methanol and Milli Q water. The samples were applied to the cartridges. The cartridges were washed with 5% methanol. Then, the cartridges were washed with 5% methanol and 2% ammonia hydroxide. The cartridges were further washed with 5% methanol and 2% formic acid. Finally, the cartridges were washed with 15% methanol and 2% formic acid. F₂-isoprostanes were eluted with hexane/ethyl acetate/2-propanol (30/65/5) solution. The solvent was evaporated to dry by a stream of nitrogen gas. The residue was dissolved in 200 μ L of 0.1% formic acid in 50/50 Milli Q water/methanol, filtered through a 0.22 μ m filter and was ready for LC–MS/MS analysis.

2.6. Instrumentation

The HPLC system was Agilent 1100 series HPLC which included G1322A vacuum degasser, G1311A binary pump, G1329A auto-sampler and G1330B thermostatic column compartment. The column was Synergi Hydro-RP 80 Å, 250 × 2.0 mm, 4 μ m column with C-18 guard column (Phenomenex, Torrance, CA). The column temperature was controlled at 25 °C. The mobile phase A was 0.1% formic acid in Milli Q water, mobile phase B was methanol. The flow rate was 150 μ L/min and the injection volume was 40 μ L. The gradient was 0–14 min 70% methanol; 14–19 min 70–95% methanol; 19–21 min 95–70% methanol; then the column was re-equilibrated to the starting condition.

The mass spectrometer was 4000 QTRAP mass spectrometer (Applied Biosystems, Foster City, CA). The ion source was Turbolon Spray (TIS). The scan mode was multiple reaction monitoring (MRM) and the polarity was negative. Source-dependent parameters: Gas 1, Gas 2, Source temperature, Curtain gas, and Ion spray voltage were set at 40 psi, 60 psi, 450 °C, 10 psi, and –4500 V, respectively. Compound-dependent parameters: declustering potential (DP), entrance potential (EP), collision cell exit potential (CXP), collision energy (CE) were set at –86, –15, –11, and –36 V, respectively. Deprotonated molecular ion [M–H][–] was selected at the first quadrupole (Q1) for standard 8-iso-PGF_{2 α} (m/z = 353) and internal standard (IS) 8-iso-PGF_{2 α} -d₄ (m/z = 357). MRM transition ion pair 353/193 was chosen for quantification. Ion pair 353/309 was the second abundant product ion, but it was not a specific ion pair for F₂-isoprostanes. For IS 8-iso-PGF_{2 α} -d₄, ion pair 357/197 was chosen for quantification. The quantification was based on the peak area ratio of the standard 8-iso-PGF_{2 α} MRM transition ion pair (353/193) to IS 8-iso-PGF_{2 α} -d₄ MRM transition ion pair (357/197). The calibration curve linear range was 0.1–50 ng/mL ($y = 0.6008x - 0.0966$, $r^2 = 0.9997$). The limit of detection (LOD) was 0.05 ng/mL and the LOQ the limit of quantification (LOQ) was 0.1 ng/mL. Analyst 1.4.2 software (Applied Biosystems) was used to control the LC–MS/MS instrument and process the data.

2.7. Statistics

Data were expressed as the mean \pm SD of three experiments. Student's *t*-test was used for significance testing, using a *p* value of 0.05.

3. Results and discussion

3.1. Nanoparticle cytotoxicity study

A549 cells were exposed to each of silica oxide and metal oxide nanoparticles for 24–48 h. Cells were treated with SiO₂ (15 nm) nanoparticle for 48 h. The dosage levels were 5, 10, 25, 50, 75, and 100 μ g/mL, with cell viability of 99.0%, 98.3%, 90.4%, 88.2%, 83.9%, and 78.3%, respectively, compared with the control cells. Among these, cell viability decreased significantly with dosage levels of 75 and 100 μ g/mL, respectively, compared with the control group ($p < 0.05$) [21]. After cells were exposed to Fe₂O₃ (30 nm) nanoparticle for 48 h at the dosage levels of 30, 40, 50, 80 and 100 μ g/mL, cell viability decreased to 93.3%, 88.9%, 83.5%, 80.5%, and 77.2%, respectively, compared with the control cells. At dosage levels of 80 and 100 μ g/mL, cell viability decreased significantly. ($p < 0.05$) [21]. ZnO nanoparticle was the most toxic among the five nanoparticles. After cells were exposed to ZnO (70 nm) nanoparticle for only 24 h, the cell viability decreased to 73.4%, 34.1% and 25.9%, respectively, at the dosage levels of 12, 14, and 16 μ g/mL ($p < 0.05$) [20]. Al₂O₃ (13 nm) nanoparticle did not show significant cytotoxicity after 48 h exposure at the dosage levels of 5, 10, 25, 50, 75, and

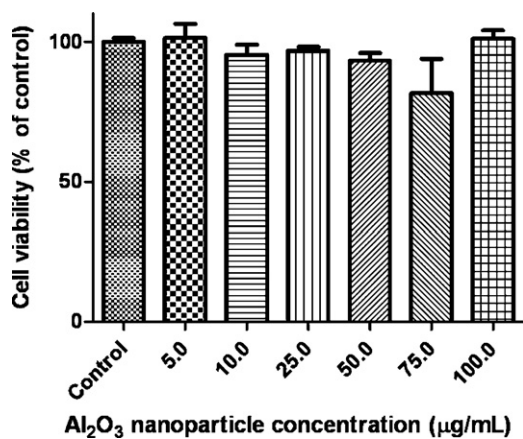


Fig. 2. A549 cell viability after 48 h exposure to 5, 10, 25, 50, 75, and 100 µg/mL of 13 nm Al₂O₃ nanoparticle. Values are mean ± SD (*n* = 3). *p* > 0.05.

100 µg/mL (Fig. 2). TiO₂ (40 nm) nanoparticle also did not show significant cytotoxicity at the dosage levels of 1, 10, 50, and 100 µg/mL after 48 h exposure (Fig. 3). The cell viability data demonstrated different degrees of cytotoxicity for SiO₂ (15 nm), Fe₂O₃ (30 nm) and ZnO (70 nm) nanoparticles, while Al₂O₃ (13 nm) and TiO₂ (40 nm) did not show significant cytotoxicity to the cells after 24–48 h exposure.

3.2. F₂-isoprostane isomers separation and quantification by LC-MS/MS

Six isomers were chosen in this study because of the commercial availability of these standards. The separation of F₂-isoprostane isomers using LC is challenging. These regioisomers have very close polarity and they are difficult to be separated on a LC column. To achieve the separation, different columns have been tested, and Synergi Hydro-RP 250 × 2 mm, 4 µm column (from Phenomenex) provided the best separation. The six isomers were separated between 8 and 13 min (as shown in Fig. 4).

Different mobile phases have also been studied and optimized. Since F₂-isoprostanes were easily deprotonated and formed negative ions, 5 mM ammonium formate in water and 5 mM ammonium formate in methanol was used at the beginning. The F₂-isoprostane isomers can be well separated, but the sensitivity was very low (data not shown). When the mobile phases were changed to 0.1% formic acid in water and methanol, the compounds gained 10 times

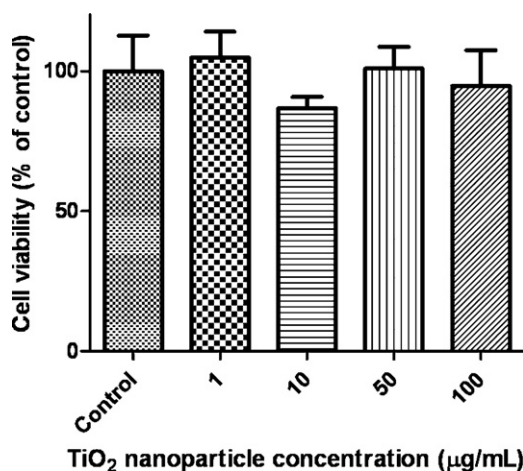


Fig. 3. A549 cell viability after 48 h exposure to 1, 10, 50, and 100 µg/mL of 40 nm TiO₂ nanoparticle. Values are mean ± SD (*n* = 3). *p* > 0.05.

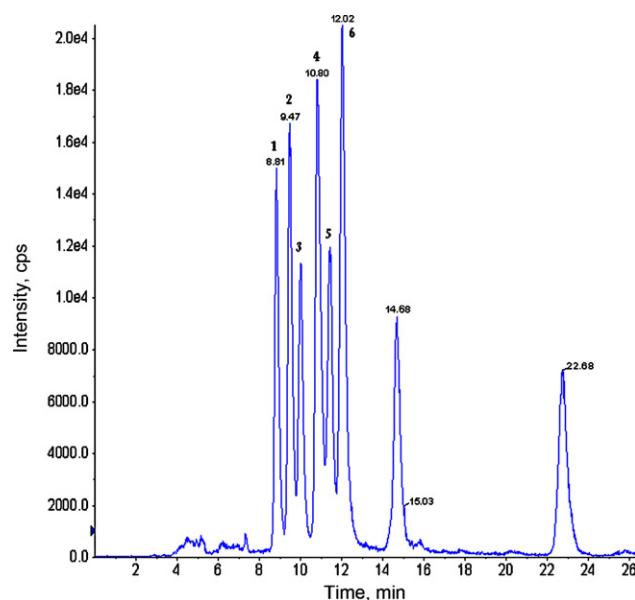


Fig. 4. The separation of F₂-isoprostane isomers. (1) 8-iso-15R-PGF_{2α}, (2) 8-iso-PGF_{2α}, (3) 11β-PGF_{2α}, (4) 15R-PGF_{2α}, and (5) 5-trans-PGF_{2α}, (6) PGF_{2α}.

sensitivity compared with 5 mM ammonium formate as buffer. It was also discovered that different organic solvents affected the F₂-isoprostanes detection sensitivity significantly. Methanol provided higher sensitivity than acetonitrile. Therefore, 0.1% formic acid in water and methanol were chosen for the mobile phase through the entire study.

MRM transition ion pair 353/193 was monitored for the isomers and ion pair 357/197 was monitored for the IS. These are consistent with the literature reports [22,23]. Extracted ion chromatograms of standard 8-iso-PGF_{2α} and IS were shown in Fig. 5. The retention times for both the standard 8-iso-PGF_{2α} and IS were 9.12 min.

3.3. Sample preparation

Sample preparation is a crucial step in the determination of F₂-isoprostanes because cell medium is a complicated biological matrix and F₂-isoprostane isomers are at trace levels in the cell. Free

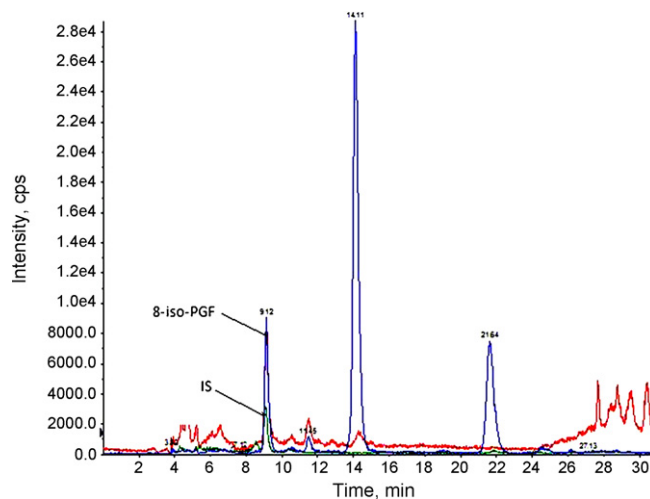


Fig. 5. Extracted ion chromatogram of standard 8-iso-PGF_{2α} and IS 8-iso-PGF_{2α}-d₄. The retention time for the standard 8-iso-PGF_{2α} and IS is 9.12 min. The blue trace is MRM ion pair 353/193. The green trace is MRM ion pair 357/197. The red trace is MRM ion pair 353/309. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of the article.)

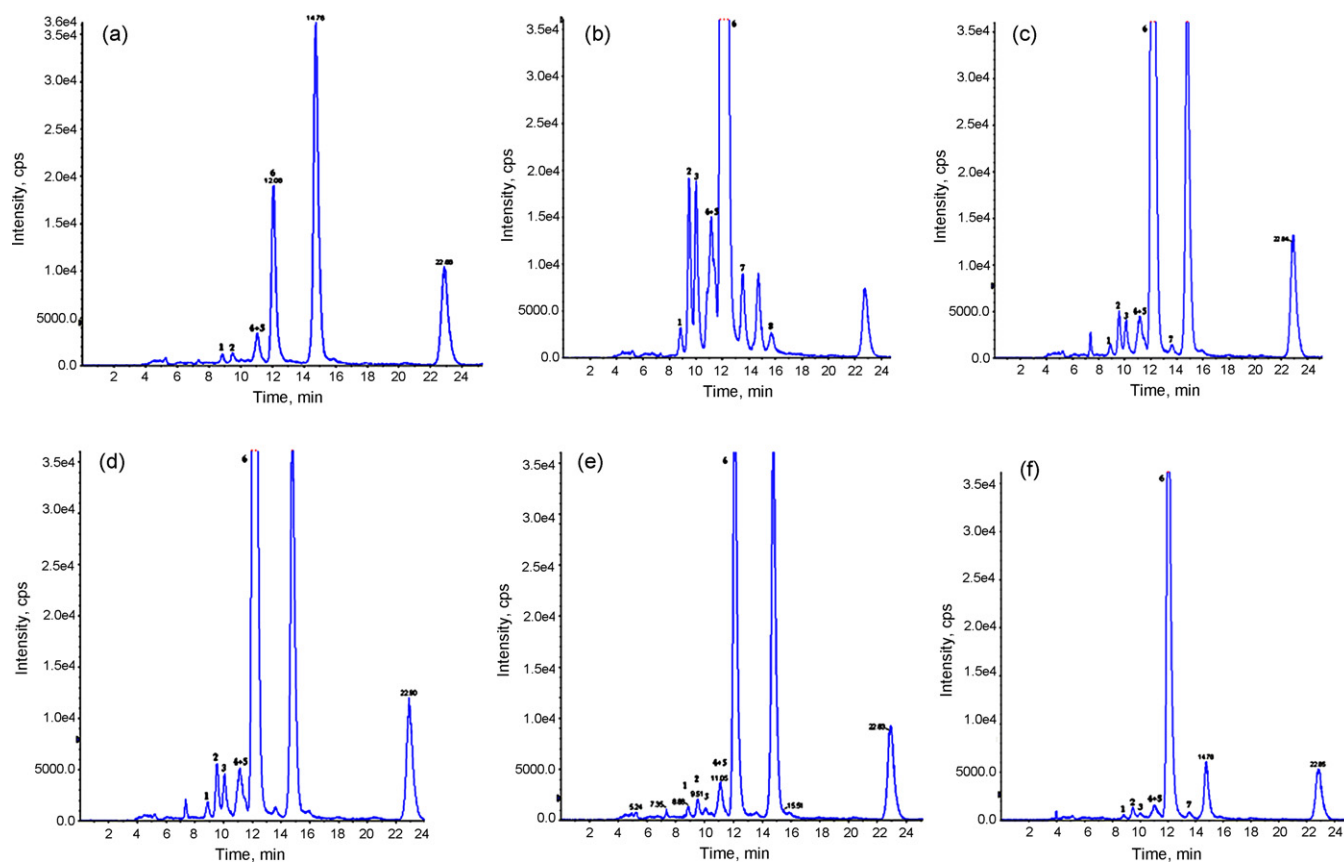


Fig. 6. The extracted ion chromatograms of F_2 -isoprostane isomers in (a) the control, (b) SiO_2 (50 $\mu\text{g/mL}$), (c) Al_2O_3 (100 $\mu\text{g/mL}$), (d) TiO_2 (100 $\mu\text{g/mL}$), (e) Fe_2O_3 (100 $\mu\text{g/mL}$) and (f) ZnO (15 $\mu\text{g/mL}$) nanoparticle-treated cells. Peaks (1) 8-iso-15R-PGF $_{2\alpha}$, (2) 8-iso-PGF $_{2\alpha}$, (3) 11 β -PGF $_{2\alpha}$, (4) + (5) 15R-PGF $_{2\alpha}$ and 5-trans-PGF $_{2\alpha}$, (6) PGF $_{2\alpha}$, (7) Unknown 1, and (8) Unknown 2.

F_2 -isoprostanes are released into the cell medium by the enzyme named as phospholipase. F_2 -isoprostanes are extracted by solid phase extraction from the cell culture medium. At first, C18 cartridge (Waters, Milford, MA) was used to extract F_2 -isoprostanes, but the recovery was only at about 20%. Oasis HLB cartridge was then used for sample extraction. After sample was loaded on the cartridge, consecutive washing with methanol, acidified water and basic water cleaned up any salted, acidic and basic compounds in the sample. At low concentration (2ng/mL), the recoveries of 8-iso-PGF $_{2\alpha}$ and 8-iso-PGF $_{2\alpha}$ -d $_4$ standards were 77.6 ± 2.4 and 77.1 ± 4.9 ($n = 3$) respectively. At high concentration (10 ng/mL), the recoveries of 8-iso-PGF $_{2\alpha}$ and 8-iso-PGF $_{2\alpha}$ -d $_4$ were 80.7 ± 2.1 and 80.5 ± 4.1 ($n = 3$) respectively. From the mass spectrum, we did not find any major peak to interfere with the F_2 -isoprostane analytes. Different elution solvents have also been studied. Methanol can elute F_2 -isoprostanes, but it caused ion suppression (date not shown). Ethyl acetate can effectively elute F_2 -isoprostanes and no ion suppression was found. This solid phase extraction procedure is only one cartridge procedure and is easier, faster and cheaper than the previous procedures requiring both HLB cartridge and μ Elution plate [30].

3.4. Levels of F_2 -isoprostane isomers in cell media after nanoparticle treatment

In this study, the A549 cells were exposed to each of five nanoparticles for 24 h. F_2 -isoprostane isomers were tentatively identified by retention time and MRM transition 353/193. The levels of these isomers were then quantified by the developed LC-MS/MS method. In the control cells, the 11 β -PGF $_{2\alpha}$

isomer's concentration was below the detection limit. Other five isomers including 8-iso-15R-PGF $_{2\alpha}$, 8-iso-PGF $_{2\alpha}$, 15R-PGF $_{2\alpha}$, 5-trans-PGF $_{2\alpha}$, and PGF $_{2\alpha}$ were detectable. 15R-PGF $_{2\alpha}$ and 5-trans-PGF $_{2\alpha}$ isomers were coeluted. The extracted ion chromatogram of F_2 -isoprostane isomers of the control is shown in Fig. 6a.

3.4.1. F_2 -isoprostane isomers in SiO_2 nanoparticle-treated cell media

After cells were exposed to SiO_2 nanoparticles (15 nm) at the dosage levels of 50 and 100 $\mu\text{g/mL}$, the levels of six F_2 -isoprostane isomers increased significantly compared with the control ($p < 0.05$) and the concentration increase was dose dependent. Moreover, two new peaks (Unknown 1 and Unknown 2) were detected next to the PGF $_{2\alpha}$ isomer after SiO_2 nanoparticle exposure. It is possible that they are new isomers of F_2 -isoprostanes which have the same mass to charge ratio. Due to the lack of standards, we cannot identify them at this time. The extracted ion chromatogram of F_2 -isoprostane isomers after SiO_2 nanoparticle (50 $\mu\text{g/mL}$) treatment is shown in Fig. 6b.

3.4.2. F_2 -isoprostane isomers in Al_2O_3 nanoparticle-treated cell media

After A549 cells were exposed to Al_2O_3 nanoparticles (13 nm) at the dosage level of 100 $\mu\text{g/mL}$, the levels of 8-iso-PGF $_{2\alpha}$, 11 β -PGF $_{2\alpha}$, 15R-PGF $_{2\alpha}$ and 5-trans-PGF $_{2\alpha}$, and PGF $_{2\alpha}$ isomers increased significantly compared with the control ($p < 0.05$). The level of 8-iso-15R-PGF $_{2\alpha}$ isomer also increased compared with the control, but was not significant ($p > 0.05$). Even though the results were similar to those of SiO_2 nanoparticle treatment, only Unknown 1 was detected in Al_2O_3 nanoparticle-treated cells. The extracted ion

Table 2
11 β -PGF_{2 α} isomer and two unknowns concentrations before and after nanoparticle treatment (mean \pm SD), $n = 3$.

Isomers	11 β -PGF _{2α} (ng/million cells)	Unknown 1 (ng/million cells)	Unknown 2 (ng/million cells)
Control	<0.0065	<0.0065	<0.0065
SiO ₂ (50 μ g/mL)	0.700 \pm 0.215*	0.773 \pm 0.527*	0.118 \pm 0.033*
SiO ₂ (100 μ g/mL)	1.87 \pm 0.496*	3.32 \pm 0.136*	0.481 \pm 0.009*
Al ₂ O ₃ (100 μ g/mL)	0.156 \pm 0.067*	0.082 \pm 0.054*	<0.0065
TiO ₂ (100 μ g/mL)	0.151 \pm 0.069*	<0.0065	<0.0065
Fe ₂ O ₃ (100 μ g/mL)	0.012 \pm 0.008*	<0.0065	<0.0065
ZnO (15 μ g/mL)	0.014 \pm 0.005*	0.061 \pm 0.031*	<0.0065

* The concentration was significantly different from that of control.

chromatogram of F₂-isoprostane isomers after Al₂O₃ nanoparticle (100 μ g/mL) treatment is shown in Fig. 6c.

3.4.3. F₂-isoprostane isomers in TiO₂ nanoparticle-treated cell media

Similarly, after A549 cells were exposed to TiO₂ (40 nm) nanoparticles at the dosage level of 100 μ g/mL, the levels of 8-iso-PGF_{2 α} , 11 β -PGF_{2 α} , 15R-PGF_{2 α} and 5-trans-PGF_{2 α} , and PGF_{2 α} isomers increased significantly compared with the control ($p < 0.05$). The level of 8-iso-15R-PGF_{2 α} isomer also increased compared with the control, but the increase was not significant ($p > 0.05$). The extracted ion chromatogram of F₂-isoprostane isomers of TiO₂ nanoparticle (100 μ g/mL) treated cells is shown in Fig. 6d.

3.4.4. F₂-isoprostane isomers in Fe₂O₃ nanoparticle-treated cell media

After A549 cells were exposed to Fe₂O₃ (30 nm) nanoparticles at the dosage level of 100 μ g/mL, the levels of 11 β -PGF_{2 α} and PGF_{2 α} isomers increased significantly compared with the control ($p < 0.05$). The levels of 8-iso-PGF_{2 α} , 15R-PGF_{2 α} and 5-trans-PGF_{2 α} isomers also increased compared with the control, but were not significant ($p > 0.05$). The level of 8-iso-15R-PGF_{2 α} isomer did not increase compared with the control. The extracted ion chromatogram of F₂-isoprostane isomers of Fe₂O₃ nanoparticle (100 μ g/mL) treated cells was shown in Fig. 6e.

3.4.5. F₂-isoprostane isomers in ZnO nanoparticle-treated cell media

After A549 cells were exposed to ZnO (70 nm) nanoparticle at the dosage level of 15 μ g/mL, the levels of 8-iso-PGF_{2 α} , 11 β -PGF_{2 α} and PGF_{2 α} isomers increased significantly compared with the control ($p < 0.05$). The level of 15R-PGF_{2 α} and 5-trans-PGF_{2 α} isomer also increased compared with the control, but were not significant ($p > 0.05$). No change was observed on the level of 8-iso-15R-PGF_{2 α} isomer. The phenomenon was very similar to those of SiO₂ (15 nm) and Al₂O₃ (13 nm) nanoparticles, and the Unknown 1 peak was detected in ZnO (70 nm) nanoparticle-treated cells. The extracted ion chromatogram of F₂-isoprostane isomers of ZnO nanoparticle (15 μ g/mL) treated cells is shown in Fig. 6f.

Total of six isomers were quantified by LC-MS/MS method in this study. In the control cells, the concentration of 8-iso-15R-PGF_{2 α} isomer was 0.0360 \pm 0.0023 ng/million cells ($n = 3$). SiO₂ (at dosage levels of 50, 100 μ g/mL), Al₂O₃, and TiO₂ nanoparticles (at the dosage levels of 100 μ g/mL) caused significant increase of 8-iso-15R-PGF_{2 α} isomer concentration compared with the control. However, Fe₂O₃ (at the dosage levels of 100 μ g/mL) and ZnO nanoparticles (at the dosage level of 15 μ g/mL) did not cause significant increasing of 8-iso-15R-PGF_{2 α} isomer concentration. The data of 8-iso-15R-PGF_{2 α} isomer after nanoparticle treatment are shown in Fig. 7.

The 8-iso-PGF_{2 α} isomer concentration in the control cell media was 0.0294 \pm 0.0007 ng/million cells ($n = 3$). After cellular exposure to SiO₂, Al₂O₃, TiO₂, Fe₂O₃, and ZnO at the dosage levels described

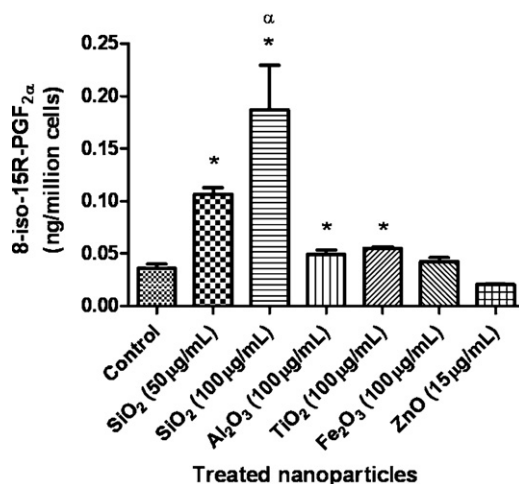


Fig. 7. 8-iso-15R-PGF_{2 α} isomer concentrations in A549 cells after SiO₂ (15 nm), Al₂O₃ (13 nm), TiO₂ (40 nm), Fe₂O₃ (30 nm) and ZnO (70 nm) nanoparticle treatment. Data with asterisk means $p < 0.05$ compared with Control. Data with α means $p < 0.05$ compared with SiO₂ nanoparticle at the concentration 50 μ g/mL.

in the last paragraph, the 8-iso-PGF_{2 α} isomer concentration in cell media increased significantly compared with the control. The detailed data of 8-iso-PGF_{2 α} isomer are shown in Fig. 8.

The 11 β -PGF_{2 α} isomer was below the detection limit in the A549 control cells. After nanoparticle exposure, the level of this isomer was detectable. All of the five nanoparticles caused a significant increase of 11 β -PGF_{2 α} isomer concentration compared with the control. The data of 11 β -PGF_{2 α} isomer after nanoparticle exposure

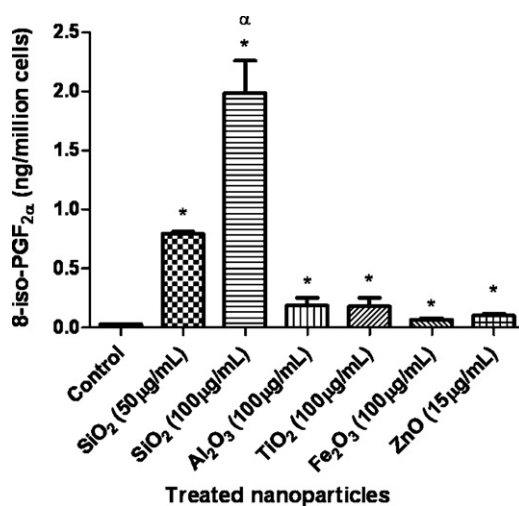


Fig. 8. 8-iso-PGF_{2 α} isomer concentrations in A549 cells after SiO₂ (15 nm), Al₂O₃ (13 nm), TiO₂ (40 nm), Fe₂O₃ (30 nm) and ZnO (70 nm) nanoparticle treatment. Data with asterisk means $p < 0.05$ compared with Control. Data with α means $p < 0.05$ compared with SiO₂ nanoparticle at the concentration 50 μ g/mL.

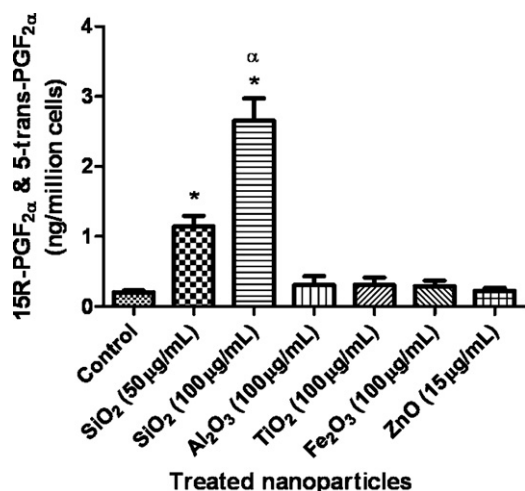


Fig. 9. 15R-PGF_{2α} and 5-trans-PGF_{2α} isomer concentrations in A549 cells after SiO₂ (15 nm), Al₂O₃ (13 nm), TiO₂ (40 nm), Fe₂O₃ (30 nm) and ZnO (70 nm) nanoparticle treatment. Data with asterisk means $p < 0.05$ compared with Control. Data with α means $p < 0.05$ compared with SiO₂ nanoparticle at the concentration 50 $\mu\text{g/mL}$.

were listed in Table 2. Since 11 β -PGF_{2α} has two *trans* ring hydroxyl groups and two *trans* alkyl chains, it is not a favorite product of free radical reaction. In addition, 5-series of isoprostanes can also form 193 product ion [31]. This deserves further study.

In the control cells, the concentrations of 15R-PGF_{2α} and 5-trans-PGF_{2α} isomers were 0.207 ± 0.015 ng/million cells ($n = 3$). After SiO₂ nanoparticle exposure at the dosage levels of 50, 100 $\mu\text{g/mL}$, the concentrations of 15R-PGF_{2α} and 5-trans-PGF_{2α} isomers increased significantly compared with the control. However, Al₂O₃, TiO₂, Fe₂O₃, and ZnO nanoparticle at the same dosage levels described above did not caused significant increase of 15R-PGF_{2α} and 5-trans-PGF_{2α} isomer concentrations. The data of 15R-PGF_{2α} and 5-trans-PGF_{2α} isomers after nanoparticle treatment are shown in Fig. 9.

Among these six isomers, PGF_{2α} was the most abundant isomer and its concentration in the control cells was 1.032 ± 0.106 ng/million cells ($n = 3$). All five nanoparticles increased its concentration significantly compared with the control. The data of PGF_{2α} isomer after nanoparticle treatment are shown in Fig. 10. PGF_{2α} is both a cyclooxygenase (COX) enzyme and free radical product, which may explain its greater abundance in the cells. Whether nanoparticles increased COX activity deserves further study.

In the SiO₂ nanoparticle exposure study, two new peaks (Unknown 1 and Unknown 2) were found compared with the control. However, in Al₂O₃ and ZnO nanoparticle exposure study, only one new peak (Unknown 1) was detected. Their data were listed in Table 2. The data implied that even though SiO₂, Al₂O₃, and ZnO can cause oxidative stress and cellular membrane damage, the detailed mechanism may not be totally the same. Moreover, 5-series of F₂-isoprostanes also can produce 193 product ion. It is possible that Unknown 1 and Unknown 2 are 5-series of F₂-isoprostanes. This phenomenon deserves further study.

Among these five types of studied nanoparticles, SiO₂ nanoparticles induced the highest increase of F₂-isoprostane isomers compared with other four metal oxide nanoparticles. Comparing with the cytotoxicity data of SiO₂, the cell viabilities were 88.2% and 78.3%, respectively, at the dosage levels of 50 and 100 $\mu\text{g/mL}$ after 48 h exposure. For ZnO (70 nm) nanoparticles, the phenomenon was very different. Our previous study showed that ZnO nanoparticle was the most cytotoxic and the cell viability was 34.1% at the dosage level of 14 $\mu\text{g/mL}$ after 24 h exposure. However, F₂-isoprostane isomers concentrations were lower in ZnO

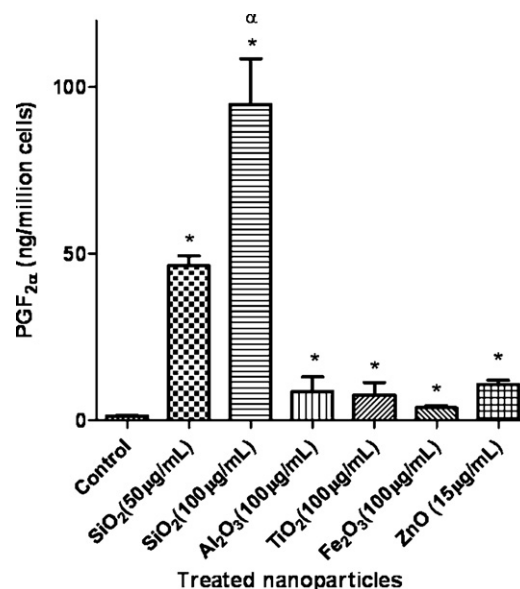


Fig. 10. PGF_{2α} isomer concentrations in A549 cells after SiO₂ (15 nm), Al₂O₃ (13 nm), TiO₂ (40 nm), Fe₂O₃ (30 nm) and ZnO (70 nm) nanoparticle treatment. Data with asterisk means $p < 0.05$ compared with Control. Data with α means $p < 0.05$ compared with SiO₂ nanoparticle at the concentration 50 $\mu\text{g/mL}$.

nanoparticle-treated cells than those of SiO₂ nanoparticle-treated cells. These data revealed that SiO₂ nanoparticles induced higher degree of lipid peroxidation and cell membrane damage than ZnO, Al₂O₃, TiO₂, and Fe₂O₃ metal oxide nanoparticles. Here, the data showed slightly different patterns of 15-series of F₂-isoprostane isomers corresponding to six nanoparticles exposure. What caused this difference deserves further study. Our previous study showed that ZnO nanoparticle can cause DNA damage [20]. Therefore, these nanoparticles may involve different mechanisms in causing oxidative stress. Our data showed that TiO₂ nanoparticles increased F₂-isoprostane isomers and cause cell membrane damage which was consistent with previous cytotoxicity study on TiO₂ nanoparticle [14]. A separate study in our research group using imaging technique showed that Al₂O₃ (13 nm) nanoparticles can cause cell membrane depolarization in A549 cells [33], which also supported our data in this study.

F₂-isoprostane isomers have been widely studied in the plasma and urine to serve as reliable biomarkers for oxidative stress [34,35]. Urinary F₂-isoprostane metabolites including 2,3-dinor-5,6-dihydro-8-iso-PGF_{2α} and 2,3-dinor-8-iso-PGF_{2α} have greater abundance than F₂-isoprostanes and also can serve as biomarkers for oxidative stress [36]. It has been reported that cigarette smokers had higher concentrations of urinary F₂-isoprostanes and their metabolites [23]. Based on our best knowledge, our study is the first report of increased F₂-isoprostane isomers in A549 cells after nanoparticle exposure. 8-Iso-PGF_{2α} isomer is considered as the most important isomers of F₂-isoprostanes because its structure is favorite to be formed by free radicals and has higher abundant than other isomers [37]. Some studies have shown that it possesses some biological functions [24–26]. In our study, PGF_{2α} was also the most abundant isomer and increased dramatically after nanoparticle treatment. Five types of studied nanoparticles can all induce significant increase of 8-iso-PGF_{2α} and PGF_{2α} isomer concentrations.

4. Conclusions

For the first time, a LC-MS/MS method has been developed to determine F₂-isoprostane isomers in human lung epithelial

cells after exposure to SiO₂ (15 nm), Al₂O₃ (13 nm), TiO₂ (40 nm), Fe₂O₃ (30 nm), and ZnO (70 nm) nanoparticles. After 24 h exposure, the concentrations of F₂-isoprostane isomers increased at different levels compared with the control cells. Data showed that these nanoparticles can induce lipid peroxidation and cell membrane damage at different degrees. SiO₂ nanoparticles induced significantly higher concentrations of F₂-isoprostane isomers than other four metal oxide nanoparticles in A549 cells. A simplified solid phase extraction procedure was also developed without ion suppression in mass spectrometer. The time response to the concentrations of biomarker F₂-isoprostane isomers and the size effects of nanoparticles on levels of the F₂-isoprostane isomers will be further studied. At the same time, the identification of two detected unknown peaks and screen for other isomers and metabolites of F₂-isoprostanes due to nanoparticle stimulation will be performed. Our data in this study will help researchers understanding the mechanism of nanotoxicity and protecting people's health when we work with nanoparticles.

Acknowledgements

The authors thank the financial support from the Department of Chemistry and Environmental Research Center at the Missouri University of Science and Technology.

References

- [1] A. Nel, T. Xia, L. Maedler, N. Li, *Science* (Washington, DC, USA) 311 (2006) 622–627.
- [2] B. Weidenhof, M. Reiser, K. Stowe, W.F. Maier, M. Kim, J. Azurdia, E. Gulari, E. Seker, A. Barks, R.M. Laine, *J. Am. Chem. Soc.* 131 (2009) 9207–9219.
- [3] R. Voggu, S. Pal, S.K. Pati, C.N.R. Rao, *J. Phys.: Condens. Matter* 20 (2008), 215211/215211–215211/215215.
- [4] Y. Xuan, X. Zhao, J. Li, Y. Liu, L. Qin, *Runhua Yu Mifeng* 34 (2009) 14–16.
- [5] Y.-T. Ke, H.-F. Wu, *Huaxue* 66 (2008) 231–240.
- [6] H. Lord, S.O. Kelley, *J. Mater. Chem.* 19 (2009) 3127–3134.
- [7] K. Tamada, *Mol. Electron. Bioelectron.* 19 (2008) 95–96.
- [8] S. Prakash, A.G. Kulamarva, *Recent Pat. Drug Delivery Formulation* 1 (2007) 214–221.
- [9] F. Zhou, D. Xing, Z. Ou, B. Wu, D.E. Resasco, W.R. Chen, *J. Biomed. Opt.* 14 (2009), 021009/021001–021009/021007.
- [10] G.J. Nohynek, E.K. Dufour, M.S. Roberts, *Skin Pharmacol. Physiol.* 21 (2008) 136–149.
- [11] G. Oberdorster, E. Oberdorster, J. Oberdorster, *Environ. Health Perspect.* 113 (2005) 823–839.
- [12] M. Auffan, W. Achouak, J. Rose, M.-A. Roncato, C. Chaneac, D.T. Waite, A. Masion, J.C. Woicik, M.R. Wiesner, J.-Y. Bottero, *Environ. Sci. Technol.* 42 (2008) 6730–6735.
- [13] K. Fujiwara, H. Suematsu, E. Kiyomiya, M. Aoki, M. Sato, N. Moritoki, *J. Environ. Health Sci. Part A: Toxic/Hazard Subst. Environ. Eng.* 43 (2008) 1167–1173.
- [14] J.-R. Gurr, A.S.S. Wang, C.-H. Chen, K.-Y. Jan, *Toxicology* 213 (2005) 66–73.
- [15] Y.-H. Hsin, C.-F. Chen, S. Huang, T.-S. Shih, P.-S. Lai, P.J. Chueh, *Toxicol. Lett.* 179 (2008) 130–139.
- [16] S.M. Hussain, K.L. Hess, J.M. Gearhart, K.T. Geiss, J.J. Schlager, *Toxicol. in Vitro* 19 (2005) 975–983.
- [17] H.L. Karlsson, P. Cronholm, J. Gustafsson, L. Moeller, *Chem. Res. Toxicol.* 21 (2008) 1726–1732.
- [18] W. Lin, Y.-w. Huang, X.-D. Zhou, Y. Ma, *Int. J. Toxicol.* 25 (2006) 451–457.
- [19] W. Lin, Y.-W. Huang, X.-D. Zhou, Y. Ma, *Toxicol. Appl. Pharmacol.* 220 (2007) 226.
- [20] W. Lin, Y. Xu, C.-C. Huang, Y. Ma, K.B. Shannon, D.-R. Chen, Y.-W. Huang, *J. Nanopart. Res.* 11 (2009) 25–39.
- [21] X. Liu, P. Whitefield, Y. Ma, *Toxicol. Environ. Chem.*, in press. doi:10.1080/02772240903127292.
- [22] H. Yin, E.S. Musiek, J.D. Morrow, *J. Biol. Sci. (Faisalabad, Pak.)* 6 (2006) 469–479.
- [23] A.W. Taylor, R.S. Bruno, M.G. Traber, *Lipids* 43 (2008) 925–936.
- [24] K. Takahashi, T.M. Nammour, M. Fukunaga, J. Ebert, J.D. Morrow, L.J. Roberts II, R.L. Hoover, K.F. Badr, *J. Clin. Invest.* 90 (1992) 136–141.
- [25] J.D. Morrow, T.A. Minton, L.J. Roberts II, *Prostaglandins* 44 (1992) 155–163.
- [26] P. Minuz, G. Andrioli, M. Degan, S. Gaino, R. Ortolani, R. Tommasoli, V. Zuliani, A. Lechi, C. Lechi, *Arterioscler. Thromb. Vasc. Biol.* 18 (1998) 1248–1256.
- [27] J.D. Morrow, L.J. Roberts II, *Methods Enzymol.* 300 (1999) 3–12.
- [28] J.D. Morrow, W.E. Zackert, J.P. Yang, E.H. Kurhts, D. Callewaert, R. Dworski, K. Kanai, D. Taber, K. Moore, J.A. Oates, L.J. Roberts, *Anal. Biochem.* 269 (1999) 326–331.
- [29] M. Haschke, Y.L. Zhang, C. Kahle, J. Klawitter, M. Korecka, L.M. Shaw, U. Christians, *Clin. Chem. (Washington, DC, USA)* 53 (2007) 489–497.
- [30] B. Zhang, K. Saku, *J. Lipid Res.* 48 (2007) 733–744.
- [31] A.W. Taylor, R.S. Bruno, B. Frei, M.G. Traber, *Anal. Biochem.* 350 (2006) 41–51.
- [32] X.D. Zhou, W. Huebner, H.U. Anderson, *Appl. Phys. Lett.* 80 (2002) 3814–3816.
- [33] W. Lin, I. Stayton, Y.-w. Huang, X.-D. Zhou, Y. Ma, *Toxicol. Environ. Chem.* 90 (2008) 983–996.
- [34] T. Sicilia, A. Mally, U. Schauer, A. Paehler, W. Voelkel, *J. Chromatogr., B: Anal. Technol. Biomed. Life Sci.* 861 (2008) 48–55.
- [35] G.L. Milne, H. Yin, J.D. Brooks, S. Sanchez, L.J. Roberts II, J.D. Morrow, *Methods Enzymol.* 433 (2007) 113–126.
- [36] Y. Liang, P. Wei, R.W. Duke, P.D. Reaven, S.M. Harman, R.G. Cutler, C.B. Heward, *Free Radical Biol. Med.* 34 (2003) 409–418.
- [37] L.J. Roberts, J.D. Morrow, *Free Radical Biol. Med.* 28 (2000) 505–513.